

Available online at www.sciencedirect.com

Tetrahedron Letters

Tetrahedron Letters 48 (2007) 8060–8064

Combinatorial synthesis of 4-oxo-4H-imidazo[1,5-a]quinoxalines and 4 -oxo- $4H$ -pyrazolo[1,5-a]quinoxalines

Julia H. Spatz,^{a,b,*} Michael Umkehrer,^a Cédric Kalinski,^a Günther Ross,^a Christoph Burdack,^a Jürgen Kolb^a and Thorsten Bach^b

> ^a Priaton GmbH, Gmunder Str. 37-37a, D-81739 München, Germany
^b Technical University Munich, Lichtenbergstr. 4, D.85747 Garching, Germany ^bTechnical University Munich, Lichtenbergstr. 4, D-85747 Garching, Germany

Received 13 July 2007; revised 8 August 2007; accepted 4 September 2007 Available online 8 September 2007

Abstract—A combinatorial synthetic route yielding imidazo $[1,5-a]$ quinoxalines and pyrazolo $[1,5-a]$ quinoxalines is described. The use of 2-fluoroaniline and 1H-imidazole-4-carboxylic acid, respectively, 1H-pyrazole-3-carboxylic acid in the Ugi-reaction (U-4CR) followed by a nucleophilic aromatic substitution (S_NAr) affords the imidazo- as well as the pyrazolo-[1,5-a]quinoxaline moiety in good yield and high diversity.

© 2007 Elsevier Ltd. All rights reserved.

Imidazoquinoxalines belong to an important class of heterocycles that are often found in biologically active and pharmacologically useful agents.^{[1](#page-2-0)} Synthetic sequences that enable the preparation of polysubstituted heterocycles have attracted considerable attention in recent years[.2,3](#page-2-0) The combinatorial synthesis of 'drug-like' compounds permits the fast preparation of compound libraries suitable for lead discovery and optimization.^{[4,5](#page-2-0)} Multi-component reactions (MCRs) are a powerful tool for the high-throughput screening strategy.^{$6,7$} One of the most important MCRs is the Ugi-reaction.[8](#page-2-0) In the Ugi four component reaction (U-4CR) an amine, an aldehyde, a carboxylic acid and an isocyanide react simultaneously to afford peptide-like structures in high diversity. This classical MCR followed by a post-condensation cyclization via nucleophilic aromatic substitution (S_NAr) leads to a new and versatile two-step

synthesis of imidazo- and pyrazolo^[1,5-a]quinoxalines. The first step of the synthesis is U -4 $\overline{C}R^{13}$ $\overline{C}R^{13}$ $\overline{C}R^{13}$ yielding products 5 (Scheme 1) as intermediates for the following cyclisation. The use of 2-fluoroaniline 1 and heterocyclic carboxylic acids 3 as bifunctional starting materials enables a subsequent nucleophilic aromatic substitution reaction in which the hydrogen-bearing nitrogen N1 of the heterocyclic carboxylic acid component acts as a nucleophile, and fluorine as a leaving group. $9-14$ We used two different carboxylic acids, 1H-imidazole-4-carboxylic acid and 1H-pyrazole-3-carboxylic acid leading to two different tricyclic title scaffolds 6.

The Ugi-reaction is generally initiated by the condensation of amine 1 with aldehyde 2 leading to an intermediate imine, which subsequently reacts with carboxylic acid 3 and isocyanide 4 to afford the desired product

Scheme 1. Combinatorial synthesis of 4-oxo-4H-imidazo[1,5-a]quinoxalines and 4-oxo-4H-pyrazolo[1,5-a]quinoxalines via U-4CR/S_NAr strategy.

Keywords: Ugi-reaction; Multi-component reaction; Imidazo-quinoxaline; Pyrazolo-quinoxaline; Nucleophilic aromatic substitution.

^{*} Corresponding author. Tel.: $+498945213080$; fax: $+4989452130822$; e-mail: spatz@priaton.de

5. Here 2,2,2-trifluoroethanol (TFE) turned out to be the best solvent for the MCR step. After completion of the MCR the TFE was removed in vacuo. Since the MCR products (4a–h and 6a–h) were generally obtained in high yields and high purities (determined by HPLC- MS ^{[15](#page-3-0)} they were used in the next step without further purification.[16](#page-3-0)

The subsequent S_NAr cyclisation requires a suitable base to enhance the nucleophilicity of the heterocyclic nitrogen N1 by deprotonation and capture of the hydrogen fluoride released during the reaction. The optimization (test system: 1H-pyrazole-3-carboxylic acid, p-anisaldehyde and t-butylisocyanide) of the S_NAr conditions (solvents, bases, reaction time and temperatures) is shown in Table 1. Conversions (Y) were determined by HPLC-MS after 20 min of reaction time under microwave (mw) irradiation or alternatively in pressure tubes at 120 °C after 19 h. The results show that Cs_2CO_3 or K_2CO_3 as a base in combination with microwave irradiation at a reaction temperature of 150 °C in DMF is the ideal method. All compounds were purified by crystallization or by column chromatography on silica gel. 17

Tables 2 and 3 show the results for the synthesized compounds 5a–h and 7a–h with specific yields for each

Table 1. S_NAr optimization

Base	Equiv	Solvent	reflux/mw	T (°C)	$Y(\%)$
KOtBu		DMF	reflux 19 h	120	84
KOtBu		DMF	mw 40 min	150	
K_2CO_3		DMF	mw 20 min	150	99
K_2CO_3		TFE	mw 40 min	100	
Cs ₂ CO ₃		DMF	reflux 40 h	105	40
Cs_2CO_3		DMF	mw 20 min	150	98

Y: crude product determined by HPLC-MS.

Table 2. Synthesized imidazo-[1,5-a]quinoxalines

Y1: yield of MCR-product (equimolar: aniline, aldehyde, carboxylic acid and isocyanide, TFE, rt), crude product determined by HPLC-MS. Y_2 : yield of S_NAr-product (5 equiv K₂CO₃, 1 equiv MCR-product, DMF, 150 °C microwave) crude product determined by HPLC-MS (isolated).

	\cdots	$-NH_2$ +	$N-NH$		OH + R_s ^{NC} U-4CR ∖ N∽NH∫ $\mathop{\mathsf{TFE}}_{\mathsf{rt}}$	ဂူ $\frac{R}{1}$ 2 R_{R_3} 'N. F Ő	S_N Ar K_2CO_3 `N⊂	R_{2} Ö ő	н R_{3}
		$\mathbf{1}$	2	$\bf 3$		R_1 6	DMF 150 °C mw	R_1 $\overline{7}$	
Entry R_1		\mathbf{R}_2	R_3	rt(h)			Y_1 (%) LC/(isol.) MCR-product Microwave (min)	Y_2 (%) LC/(isol.)	S_N Ar-product
$\mathbf{1}$	$\boldsymbol{\mathrm{H}}$			$16\,$	80 (77)	6a	20	94 (89)	${\bf 7a}$
\overline{c}	$\mathbf H$			16	83	6b	$20\,$	77 (61)	$7\mathrm{b}$
\mathfrak{Z}	$\mathbf H$			$16\,$	$77\,$	6c	$20\,$	65 (51)	$7\mathrm{c}$
4	$\, {\rm H}$	$\, {\rm H}$		16	89	6d	$20\,$	65 (51)	$7\mathbf{d}$
5	$\mathbf H$	$\boldsymbol{\mathrm{H}}$		$16\,$	54	6e	$20\,$	40 (37)	$7\mathrm{e}$
6	$\boldsymbol{\mathrm{H}}$			$16\,$	90	6f	$20\,$	85 (79)	$7\mathbf{f}$
τ	$3-CH_3$			$16\,$	82	6g	$40\,$	61(53)	$7\mathrm{g}$
$\,8\,$	$3-CF_3$			$16\,$	74	6h	$40\,$	77 (75)	$7\mathrm{h}$

Table 3. Synthesized pyrazolo- $[1,5-a]$ quinoxalines

 R_{1}

Y1: yield of MCR-product (equimolar: aniline, aldehyde, carboxylic acid and isocyanide, TFE, rt), crude product determined by HPLC-MS. Y_2 : yield of S_NAr-product (5 equiv K₂CO₃, 1 equiv MCR-product, DMF, 150 °C microwave), crude product determined by HPLC-MS (isolated).

step ($Y_1 = MCR$, $Y_2 = S_NAr$). Aliphatic, phenylic and benzylic isocyanides as well as aliphatic and aromatic aldehydes could be involved in the reaction successfully.

The reaction times (rt) for the S_N Ar are generally short and the conversions are good for all compounds. Chromatographic methods allow the isolation of products with high purity $(>\frac{95}{6})$. The protocol is quite robust and tolerates a broad range of starting materials.

MCR products 4a and 6a were isolated and analyzed by NMR as exemplary compounds. The spectra were compared with those of their cyclized analogs 5a and 7a. As expected the S_NAr cyclization leads to significant changes, the most prominent ones being the absence of the heterocyclic NH signals as well as the H–F coupling (aniline part) and a considerable downfield shift of the heterocyclic CH as well as the α -CH (previous aldehyde CH) signals.

In summary, a novel two-step synthesis procedure for the preparation of highly substituted 4-oxo-4H-imidazo- $[1,5-a]$ quinoxalines and 4-oxo-4H-pyrazolo $[1,5-a]$ quinoxalines has been described. Amines and carbonyls can be varied broadly, leading to compounds with three potential points of diversity.

References and notes

- 1. Chen, P.; Barrish, J. C.; Iwanowicz, E.; Lin, J.; Bednarz, M. S.; Chen, B. Tetrahedron Lett. 2001, 42, 4293.
- 2. Hermkens, P. H. H.; Ottenheijm, H. C. J.; Rees, D. C. Tetrahedron 1997, 53, 5643.
- 3. Gordon, E. M.; Gallop, M. A.; Patel, D. V. Acc. Chem. Res. 1996, 29, 144.
- 4. Gallop, M. A.; Barrett, R. W.; Dower, W. J.; Fodor, S. P. A.; Gordon, E. M. J. Med. Chem. 1994, 37, 1233.
- 5. Gordon, E. M.; Barrett, R. W.; Dower, W. J.; Fodor, S. P. A.; Gallop, M. A. J. Med. Chem. 1994, 37, 1385.
- 6. Dömling, A. Comb. Chem. High Throughput Screening 1998, 1, 1.
- 7. Dömling, A. Chem. Rev. 2006, 106, 17.
- 8. Ugi, I.; Meyer, R.; Fetzer, U.; Steinbrückner, C. Angew. Chem. 1959, 71, 386.
- 9. Wright, D. L.; Robotham, C. V.; Aboud, K. Tetrahedron Lett. 2002, 43, 943.
- 10. Xiang, Z.; Luo, T.; Cui, J.; Shi, X.; Fathi, R.; Chen, J.; Yang, Z. Org. Lett. 2004, 18, 3155.
- 11. Pirrung, M. C.; Das Sarma, K. J. Am. Chem. Soc. 2004, 126, 444.
- 12. Gedey, S.; Van der Eycken, J.; Fulop, F. Org. Lett. 2002, 11, 1967.
- 13. Kalinski, C.; Umkehrer, M.; Gonnard, S.; Jäger, N.; Ross, G.; Hiller, W. Tetrahedron Lett. 2006, 47, 2041.
- 14. (a) Tempest, P.; Ma, V.; Kelly, M. G.; Jones, W.; Hulme, C. Tetrahedron Lett. 2001, 42, 4963; (b) Tempest, P.; Pettus, L.; Gore, V.; Hulme, C. Tetrahedron Lett. 2002, 44, 1947.
- 15. HPLC-MS/MS spectra (Varian 1200), Polaris, RP C18 column, $3 \text{ mm} \times 150 \text{ mm}$, 5 µm , ProStar 320 (254 nm), 1 mL/min, 3 min gradient from 90% H_2O to 10% H_2O $(0.1\%$ HCOOH) versus CH₃CN, coupled with a Quadrupol MS/MS mass spectrometer using electrospray ionisation (ESI).
- 16. General procedure for the synthesis of MCR products $4a-h$ and 6a–h (GP-MCR): Fluoroaniline 1 (1 mmol) and aldehyde 2 (1 mmol) were stirred in 3 mL TFE for 2 h. Then, carboxylic acid (1 mmol) and isocyanide 3 (1 mmol) were added and the reaction mixture was stirred for 16 h at room temperature. The solvent was removed in vacuo. The MCR products were generally obtained in high yields and high purities (determined by HPLC-MS) and they were used in the next step without further purification.
- 17. General Procedure for the synthesis of 4-oxo-4H-imi $dazo[1,5-a]$ quinoxalines $5a-h$ and $4-oxo-4H-pvrazolo-[1,5-d]$ a]quinoxalines $7a-h$ (GP-cyclisation): (1.0 mmol) of MCRproduct 4a–h, respectively, 6a–h was dissolved in 4 mL DMF (dry) and 5 mmol K_2CO_3 was added. The reaction was stirred at 150° C for $20-40$ min under microwave irradiation (microwave system: Discover, BenchMate, CEM). Then, 10 mL water was added and the mixture was extracted with 3×15 mL of ethyl acetate. The organic layer was dried over $MgSO₄$ and the resulting crude product was purified by flash chromatography or crystallization.

Compound 4a was prepared according to GP-MCR and the resulting crude product was purified by chromatography on silica gel with eluent ethyl acetate/hexane $= 2/1$ (219 mg, 61%). $m/z = 359$ [M+H]⁺, $m/z = 381$ [M+Na]⁺. ¹H NMR (DMSO- d_6 , 250.13 MHz): 12.77 (bs, 1H), 7.97 $(s, 1H), 7.61-7.19$ (m, 6H), 4.08 (d, 1H, $J = 10.4$ Hz), 1.27 (s, 9H), 0.75 (s, 1H), 0.45–0.40 (m, 2H), 0.17 (m, 2H).

Compound 6a was prepared according to GP-MCR and the resulting crude product was purified by flash chromatography on silica gel with eluent ethyl acetate/hexane = 2/1 (259 mg, 72%). $m/z = 359$ [M+H]⁺, $m/z = 381$ $[M+Na]^{+}$. ¹H NMR (DMSO-d₆, 250.13 MHz): 13.30 (bs, 1H), 7.94 (s, 1H), 7.58–7.19 (m, 6H), 4.13 (d, 1H, $J = 10.1$ Hz), 1.51 (s, 9H), 0.79–0.74 (m, 1H), 0.49–0.38 (m, 2H), 0.21–0.08 (m, 2H).

Compound 5a was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (314 mg, 92%). $m/z = 339$ [M+H]⁺, $m/z = 361$ [M+Na]⁺. ¹H NMR (DMSO-d6, 250.13 MHz): 9.15 (s, 1H), 8.32 (d, 1H, $J = 7.4$ Hz), 7.95 (s, 1H), 7.47–7.19 (m, 3H), 7.19 (s, 1H), 5.11 (d, 1H, $J = 9.5$ Hz), 1.27 (s, 9H), 1.85–1.78 (m, 1H), 0.86–0.64 (m, 2H), 0.40–0.24 (m, 2H).

Compound 5b was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (239 mg, 61%). $m/z = 389$ [M+H]⁺, $m/z = 411$ [M+Na]⁺. ¹H NMR (DMSO-d6, 250.13 MHz): 9.52 (s, 1H), 9.19 (bs, 1H), 8.35 $(d, 1H, J = 7.7 Hz)$, 7.99 (bs, 1H), 7.56–7.44 (m, 2H), 7.38 (d, 3H, $J = 9.0$ Hz), 6.88 (d, 2H, $J = 9.0$ Hz), 5.37 (d, 1H, $J = 8.7$ Hz), 3.73 (s, 3H), 1.91–1.88 (m, 1H), 0.90–0.73 (m, 2H), 0.39–0.26 (m, 2H).

Compound 5c was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (222 mg, 55%). $m/z = 403$ [M+H]⁺; $m/z = 445$ [M+Na]⁺. ¹H NMR (DMSO-d6, 250.13 MHz): 9.12 (s, 1H), 8.44 (bs, 1H), 8.31–8.27 (m, 1H), 7.93 (s, 1 H), 7.36–7.34 (m, 3 H), 7.06 (d, 2H, 8.4 Hz), 6.79 (d, 2H, 8.5 Hz), 5.21 (m, 1H), 4.12 (d, 2H, $J = 5.9$ Hz), 3.710 (s, 3H), 1.90-1.70 (m, 1H), 0.84-0.65 (m, 2H), 0.36–0.15 (m, 2H).

Compound 5d was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (157 mg, 53%). $m/z = 297$ $[M+H]^+$, $m/z = 319$ $[M+Na]^+$. ¹H NMR (DMSO-d6, 250.13 MHz): 9.16 (s, 1H), 8.33 (d, 1H, $J = 7.7$ Hz), 7.96–7.89 (m, 2H), 7.40 (m, 3H), 5.15 (bs, 1H), 2.61 (d, 3H, $J = 4.4$ Hz), 1.82–1.78 (m, 1H), 0.87– 0.66 (m, 2H), 0.39–0.16 (m, 2H).

Compound 5e was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (261 mg, 72%). $m/z = 363$ [M+H]⁺, $m/z = 385$ [M+Na]⁺. ¹H NMR (DMSO-d6, 250.13 MHz): 9.15 (s, 1H), 8.71 (t, 1H, $J = 5.8$ Hz), 8.31 (d, 1H, $J = 6.6$ Hz), 7.91 (s, 1H), 7.51– 7.30 (m, 3H), 7.18 (d, 2H, $J = 8.7$ Hz), 6.89 (d, 2H, $J = 8.7$ Hz), 4.93 (s, 2H), 4.25 (d, 2H, $J = 5.7$ Hz), 3.76 (s, 3H).

Compound 5f was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (145 mg, 47%). $m/z = 299 \text{ [M+H]}^+, m/z = 321 \text{ [M+Na]}^+.$ ¹H NMR (DMSO-d6, 250.13 MHz): 9.16 (s, 1H), 8.31 (d, 1H, $J = 7$ Hz), 8.00 (s, 1H), 7.96 (bs, 1H) 7.27–7.49 (m, 3H), 4.84 (s, 2H), 1.30 (s, 9H).

Compound 5g was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (260 mg, 65%). $m/z = 405$ [M+H]⁺, $m/z = 427$ [M+Na]⁺. ¹H NMR (DMSO-d6, 250.13 MHz): 9.16 (s, 1H,), 8.27 (d, 1H, $J = 7.1$ Hz), 8.01 (s, 1H), 7.75 (s, 1H), 7.35–7.24 (m, 5H), 6.92 (m, 3H), 3.75 (s, 3H), 1.28 (s, 9H).

Compound 5h was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (151 mg, 37%). $m/z = 407$ $[M+H]^+$, $m/z = 429$ $[M+Na]^+$. ¹H NMR (DMSO-d6, 250.13 MHz): 9.21 (s, 1H), 8.51 (d, 1H, $J = 8.4$ Hz), 7.99 (s, 1H), 7.73 (d, 1H, $J = 8.6$ Hz), 7.68 $(s, 1H), 7.32 (s, 1H), 5.18 (d, 1H, J = 9.5 Hz), 1.70-1.65$ (m, 1H), 1.20 (s, 9H), 0.85–0.82 (m, 1H), 0.67–0.63 (m, 1H), 0.37–0.34 (m, 1H), 0.28–0.22 (m, 1H). 13C NMR (DMSO-d6, 62.89 MHz): 168.32, 155.78, 134.79, 133.19, 129.76, 127.80 (d, $\frac{2J(C,F)}{23.7 \text{ Hz}}$), 125.13, 124.50 $(d, {}^{1}J(C,F) = 272.47 \text{ Hz})$, $J^3J(C,F) = 272.47 \text{ Hz}$, 122.43, 120.95 (d, $J^3J(C,F) = 3.6 \text{ Hz}$), 118.57, 114.73 (d, $J^3J(C,F) = 3.6 \text{ Hz}$), 99.99, 60.09, 51.88, 28.98, 12.18, 7.60, 3.19.

Compound 7a was prepared according to GP-cyclisation and purified by chromatography on silica gel with eluent chloroform/ethyl acetate = $1/1$ (303 mg, 89%). $m/z = 339$ $[M+H]^+, m/z = 361 [M+Na]^+.$ ¹H NMR (DMSO- d_6) 250.13 MHz): 8.32 (d, 1H, $J = 7.4$ Hz), 8.15 (s, 1H), 7.55–7.43 (m, 3H), 7.25 (s, 2H), 5.15 (d, 1H, $J = 9.5$ Hz) 1.87–1.83 (m, 1H), 1.27 (s, 9H), 0.86–0.68 (m, 2H), 0.37– 0.24 (m, 2H).

Compound 7b was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (246 mg, 61%). $m/z = 403$ [M+H]⁺, $m/z = 445.0$ [M+Na]⁺. ¹H NMR (DMSO-d₆, 250.13 MHz): 8.49 (s, 1H), 8.34–8.16 (m, 1H), 8.15 (s, 1H), 7.42–7.46 (m, 3H), 7.25 (d, 1H, $J = 2.1$ Hz), 7.10 (d, 2H, $J = 8.4$ Hz), 6.83 (d, 2H, $J = 7.5$ Hz), 5.29 (bs, 1H), 4.24 (d, 2H; $J = 5.9$ Hz), 3.74 (s, 3H), 1.87–1.86 (m, 1H), 0.89–0.72 (m, 2H), 0.41–0.19 (m, 2H).

Compound 7c was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (151 mg, 51%). $m/z = 297$ [M+H]⁺, $m/z = 319$ [M+Na]⁺. ¹H NMR $(DMSO-d_6, 250.13 MHz)$: 8.30 (d, 1H, $J = 5.6$ Hz), 8.12 $(d, 1H, J = 2.0 \text{ Hz})$, 7.89–7.88 (m, 1H), 7.46–7.38 (m, 3H), 7.22 (d, 2H, $J = 2.0$ Hz), 5.16 (m, 1H), 2.59 (d, 3H,

 $J = 4.4$ Hz), 1.83–1.78 (m, 1H), 0.88–0.38 (m, 2H), 0.36– 0.14 (m, 2H).

Compound 7d was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (190 mg, 51%). $m/z = 363 \left[\text{M} + \text{H} \right]^+$, $m/z = 385 \left[\text{M} + \text{Na} \right]^+$. ¹H NMR $(DMSO-d_6, 250.13 MHz)$: 8.70 (t, 1H, $J = 4.4$ Hz), 8.30 (d, 1H, $J = 8.1$ Hz), 8.15 (d, 1H, $J = 2.1$ Hz), 7.49–7.41 (m, 3H), 7.25 (d, 1H, $J = 2.1$ Hz), 7.18 (d, 2H, $J = 8.7$ Hz), 6.90 (d, 2H, $J = 8.7$ Hz), 5.00 (s, 2H), 4.26 (d, 2H, $J = 5.7$ Hz), 3.76 (s, 3H); ¹³C NMR (DMSO-d₆, 62.89 MHz): 167.28, 159.15, 155.04, 142.64, 132.65, 131.88, 130.01, 129.36, 127.98, 125.34, 124.74, 116.93, 116.11, 114.57, 108.14, 55.98, 45.30, 42.55.

Compound 7e was prepared according to GP-cyclisation and purified by crystallization from dietyhl ether (124 mg, 37%). $m/z = 333[M+H]^+$, $m/z = 355 [M+Na]^+$. ¹H NMR (DMSO-d6, 250.13 MHz): 10.34 (s, 1H), 8.29 (d, 1H, $J = 9.0$ Hz), 8.14 (d, 1H, $J = 1.9$ Hz), 7.57–7.39 (m, 5H), 7.24 (d, 1H, $J = 2.0$ Hz), 7.12 (d, 2H, $J = 8.2$ Hz), 5.15 (s, 2H), 2.25 (s, 3H).

Compound 7f was prepared according to GP-cyclisation and the resulting crude product was purified by flash chromatography on silica gel with eluent chloroform $(320 \text{ mg}, 79\%)$. $m/z = 405 \text{ [M+H]}^+, m/z = 427 \text{ [M+Na]}^+.$ H NMR (DMSO-d6, 250.13 MHz): 8.29 (d, 1H, $J = 7.9$ Hz), 8.16 (d, 1H, $J = 2.1$ Hz), 7.78 (s, 1H), 7.44– 7.28 (m, 6H), 6.93 (d, 2H, $J = 1.7$ Hz), 6.90 (s, 1H), 3.75 (s, 3H), 1.29 (s, 9H).

Compound 7g was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (222 mg, 53%). $m/z = 419$ [M+H]⁺, $m/z = 441$ [M+Na]⁺. ¹H NMR $(DMSO-d_6, 250.13 MHz): 8.15$ (d, 1H, $J = 8.2$ Hz), 8.10 (d, 1H, $J = 2.0$ Hz), 7.59 (s, 1H), 7.26–7.15 (m, 5H), 6.88 $(d, 2H, J = 8.7 \text{ Hz})$, 6.80 (s, 1H), 3.72 (s, 3H), 2.24 (s, 3H), 1.23 (s, 9H).

Compound 7h was prepared according to GP-cyclisation and purified by crystallization from diethyl ether (306 mg, 75%). $m/z = 407$ [M+H]⁺, $m/z = 429$ $(306 \text{ mg}, 75\%).$
 $[M+Na]^{+}.$ ¹H H NMR (DMSO- d_6 , 250.13 MHz): 8.48 (d, 1H, $J = 8.9$ Hz), 8.24 (d, 1H, $J = 1.7$ Hz), 7.79 (s, 2H), 7.41 (s, 1H), 7.34 (d, 1H, $J = 1.7$ Hz), 5.25 (d, 1H, $J = 9.5$ Hz), 1.76–1.72 (m, 1H), 1.24 (s, 9H), 0.87–0.85 (m, 1H), 0.74–070 (m, 1H), 0.40–0.26 (m, 2H).